

7B1 Cells, tissues, organs & systems



Life processes

- All living things have 7 life processes (MRS GREN)
- M = Movement
- R = Reproduction
- S = Sensitivity
- G = Growth
- R = Respiration
- E = Excretion
- N = Nutrition

Function of cell parts

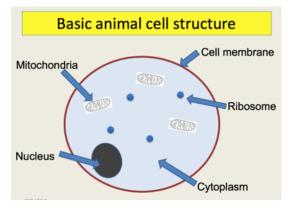
- Nucleus controls the cell, contains DNA
- Cell membrane allows substances in and out
- Cell wall shape, structure, support
- Mitochondria respiration
- Ribosomes make protein
- Vacuole (plant only) stores water and sap
- Chloroplast (plant only) absorbs sunlight energy for photosynthesis
- Cytoplasm liquid that fills the cell, chemical reactions occur here

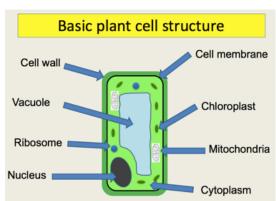
Tissues, organs, organ systems

- A tissue is a group of cells working together Eg. Muscle tissue, xylem tissue
- An organ is a group of tissues working together Eg. Heart, stomach, leaf
- An organ system is a collection of organs working together Eg. Digestive system

Magnification

Magnification = eyepiece lens x objective lens







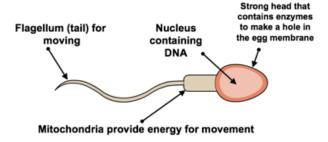
7B2 Sexual Reproduction



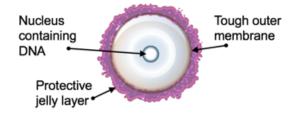
Gametes

- Gametes are sex cells
- Male gamete → sperm. Made in the testes
- Female gamete → egg. Made in the ovary

Adaptations of a sperm cell



Adaptations of an egg cell



Sexual organs

- Male → penis
- Female → vagina

Fertilisation

- Fertilisation occurs when a sperm cell fuses with an egg cell
- This happens in the oviduct
- Once sperm and egg have joined, the resulting cell is called a zygote
- The zygote multiplies to form a ball of cells called an embryo
- The embryo implants in the wall of the uterus
- This grows and becomes a foetus

Pregnancy & Birth

- Pregnancy in humans lasts for 40 weeks
- Oxygen and nutrients are passed into the foetus through the umbilical cord and placenta
- Carbon dioxide and waste products are removed from the foetus through the umbilical cord and placenta



7B3 Muscles & Bones



Muscles

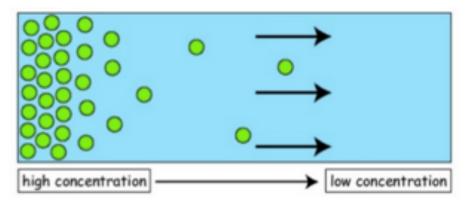
- Can only pull
- Work in antagonistic pairs eg. biceps and triceps
- Tendons attach muscles to bones

The skeleton

The four functions of the skeleton are support, protection, storage and movement

The lungs and breathing

- The lungs have a large surface area
- The diaphragm is a muscle under the lungs
- When the diaphragm contracts we inhale
- When the diaphragm relaxes we exhale
- · Gases are exchanged in the lungs by diffusion
- Diffusion is the movement of particles from a high concentration to a low concentration



The heart

- The heart is made of cardiac muscle
- The heart has four chambers

The blood

- Red blood cells carry oxygen
- White blood cells fight pathogens (germs)
- Platelets causes clots
- Plasma is the liquid part of the blood

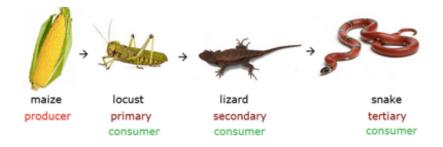


7B4 Ecosystems



Food chains

- Always start with a producer which makes its own food by by photosynthesis
- Producer → primary consumer → secondary consumer → tertiary consumer



Species

A species is a group of living things that can interbreed (have babies)

Biodiversity

Biodiversiy is the amount of different species living in a habitat

Sampling techniques - Quadrats

- Quadrat hollow square frame
- · Place randomly in the area you wish to sample
- Count up the number of organisms
- · Repeat many times and find the average
- Multiply the average by the number of quadrats that will fit into the area





7B5 Food & Nutrition

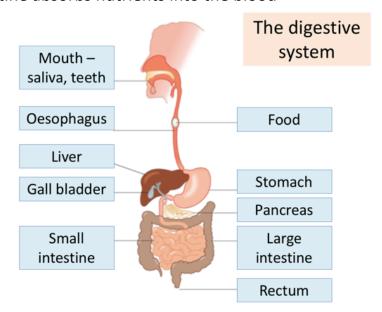


Seven food groups

- Carbohydrate used for energy
- Protein
- Fats
- Vitamins
- Minerals
- Fibre
- Water

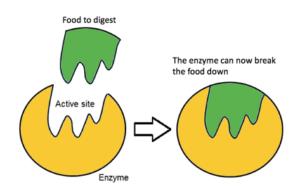
Digestive system

- Digestion is breaking large insoluble molecules into smaller, soluble ones
- The stomach contains hydrochloric acid which kills germs
- The liver produces bile
- Bile helps digest fats
- The small intestine absorbs nutrients into the blood



Enzymes

- Enzymes speed up reactions
- Enzymes work by the lock and key method





7C1 Atoms, Elements & Particles



Atoms, elements, molecules, compound

- Atoms make up everything
- An element is made up of only one type of atom
- A molecule is made up of two or more atoms
- A compound is made up of two or more different atoms
- A mixture is a group of chemicals not bonded together



The Periodic Table

- All elements have their own symbol eg. H is hydrogen, Na is sodium
- Metals are on the left and in the middle
- Non-metals are on the right

Solid, liquid and gas particle diagrams and properties

Solid	Liquid	Gas
	2823	• •
Cannot be compressed	Cannot be compressed	Can be compressed
Fixed shape Doesn't flow	Takes shape of container Flows	Takes shape of container Flows
High density	Quite high density	Low density



7C2 Mixtures & Separation

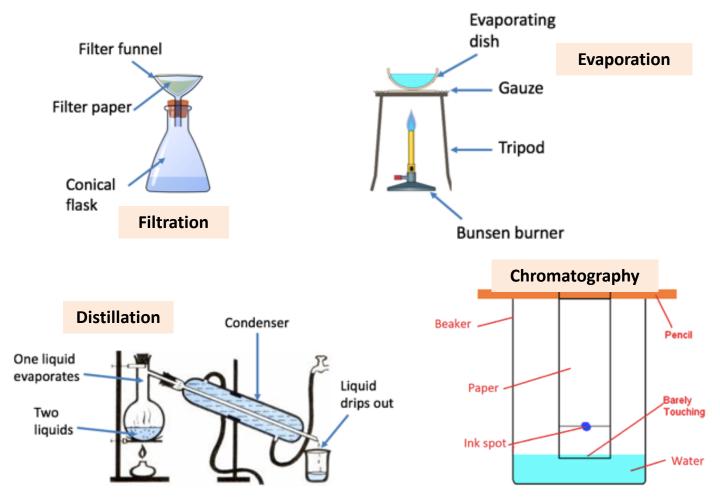


Key words

- A solute is a substance that dissolves in water eg. salt
- A solvent is a substance that dissolves other things eg. water
- A solution is a solute dissolved in a solvent eg. salty water
- Insoluble substances do not dissolve in water eg. sand
- Soluble substances do dissolve in water eg. salt
- A <u>mixture</u> is a group of chemicals not bonded together eg. air

Separating methods

- Filtration is used to separate insoluble substances from a liquid
- Evaporation is used to separate a liquid from a solid
- Distillation is used to separate liquids with different boiling points
- Chromatography is used to separate dyes of different solubility





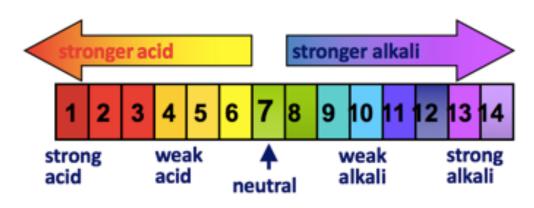
7C3 Acids & Alkalis



pH scale

- 1 to 6 is acid
- 7 is neutral
- 8 to 14 is alkali

Universal indicator colours



Key equation

Acid + alkali → salt + water

Common acids

- Hydrochloric acid
- Sulfuric acid
- Nitric acid

Common alkalis

- Sodium hydroxide
- Magnesium hydroxide
- Calcium hydroxide

Naming salts

- Hydro<u>chloric</u> acid makes metal <u>chlorides</u>
 Hydrochloric acid + sodium hydroxide → sodium chloride + water
- <u>Sulf</u>uric acid makes metal <u>sulfides</u>
 Sulfuric acid + magnesium hydroxide → magnesium sulfate + water
- <u>Nitric acid makes metal nitrates</u>
 Nitric acid + calcium hydroxide → calcium nitrate + water

Bases and alkalis

- A base is a substance that neutralises an acid
- An alkali is a soluble base (dissolves in water)



7C4 Rates of Reaction



Reversible + irreversible reactions

- Reversible reactions can be changed back eg. melting ice
- Reversible reactions are physical changes
- Irreversible reactions cannot be changed back eg. cooking an egg
- Irreversible reactions are chemical changes

Particles and reactions

- Particles are always moving
- They collide
- If they collide with enough energy there will be a reaction

Rate of reaction

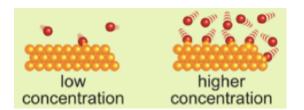
Rate of reaction is how fast the reaction goes in a set amount of time

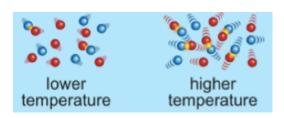
Factors that affect rate of reaction

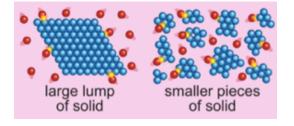
- Surface area of the chemicals. A larger surface area = faster reaction
- Concentration of chemicals. Higher concentration = faster reaction
- Temperature of chemicals. Higher temperature = faster reaction
- Catalysts speed up a reaction

Variables

- Independent variable this is what you change
- Dependent variable this is what you measure
- Control variable this is what you keep the same









7P1 Energy



Conservation of energy

Energy cannot be created or destroyed, only transferred from one form to another

Nine types of energy

- Thermal / heat
- Light
- Sound
- Chemical all fuels contain chemical energy
- Kinetic things moving
- Gravitational potential
- Elastic potential
- Nuclear
- Electrical

Fuels

- A fuel is a substance that is burned to release energy
- Energy is measured in joules (J)

Renewable energy

- Renewable energy sources will not run out
- Solar, wind, wave, hydroelectricity, tidal, geothermal, biomass

Non-renewable energy

- Non-renewable energy sources will run out
- Fossil fuels coal, oil, gas
- Nuclear



7P2 Electricity



Current

- Current is the flow of electrons
- Current is measured in amps (A)
- Current is measured with an ammeter

Voltage

- Voltage is the push given to the electrons by a battery
- Voltage is measured in volts (V)
- Voltage is measured with a voltmeter

Resistance

- Resistance means how hard it is for the current to flow
- Resistance is measured in ohms (Ω)

Series circuits

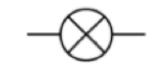
- Current is the same at all points
- Voltage is shared by the parts of the circuit

Parallel circuits

- Current splits at a junction
- Voltage is the same across all parts of the circuit

Series

Circuit symbols



Lamp (bulb)



Cell

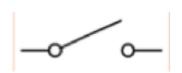
Battery



Ammeter



Voltmeter



Switch



7P3 Forces



Forces

- Force is measured in newtons (N)
- A force can change an object's shape, speed or direction

Contact Forces

- Objects need to be touching to feel a force
- Friction
- Upthrust
- Air resistance

Non-contact Forces

- Objects do not need to be touching to feel a force
- Gravity
- Electrostatic
- Magnetism

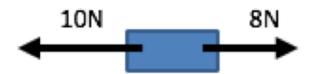
Balanced forces

Forces acting in opposite directions are the same



Non-balanced forces

- Forces acting in opposite directions are not the same
- The object will move in the direction of the largest force



The overall force is called the resultant force



7P4 Light & Sound



Waves

Waves transfer energy

Light

- Light is transverse wave
- Light travels straight lines
- Light travels faster than sound

The law of reflection

• Angle of incidence = angle of reflection

Refraction

- Refraction means light waves changing direction
- This is because light waves travel at different speeds in different objects eg. air, water, glass

Sound

- Sound is a longitudinal wave
- Sound is caused by vibrations
- Sound is slower than light
- Sound reflections are called echoes

Speed

- The speed of a wave is how fast the wave is moving
- Speed is calculated with this equation:

IInits

- Speed = metres per second (m/s)
- Distance = metres (m)
- Tme = seconds (s)